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Synthesis of 1-deoxyhept-2-ulosyl-glycono-1,5-lactone utilizing α-selective O-glycosidation of 2,6-anhydro-1-deoxy-D-hept-1-enitols

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Abstract—A series of 1-deoxy-heptulo-2-pyranosyl-glycono-1,5-lactones were synthesized utilizing completely α -selective O-glycosidation of heptenitols. Anomeric configuration of the products was confirmed by ${}^{3}J_{C,H}$ coupling measurement and X-ray crystal structural analysis. The benzyl-protected ketosyl saccharides were partly unstable, and glycosidic linkage was prone to cleave under the usual debenzylation conditions. To prevent this, we surveyed various additives for the Pd-catalyzed hydrogenation reaction and found that basic alumina was the most effective.

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1. Introduction

Bioactive sugar chains containing ketoses such as fructose and sialic acids are abundant in nature. The sialic acid family, represented by N-acetylneuraminic acid (Neu5Ac), consists of oligosaccharides or glycoconjugates in the form of α -ketosides. To take a notable example, 3-deoxy-D-manno-oct-2ulosonic acid is a constituent of glycolipids of Gram-negative bacteria and is α-linked to lipid A. Since sialic acids are involved in numerous biological processes, they are attractive targets for drug discovery.¹ It is also known that 1-deoxy- α -D-gluco-heptulose 2-phosphate, which is a 1-C-methyl analogue of D-glucose 1-phosphate, is a potent phosphorylase inhibitor. The interaction of heptulose 2-phosphate with enzymes has been examined thoroughly.² It is expected that heptulosyloligosaccharides can be utilized as a new class of bioactive oligosaccharides. In the synthetic field, therefore, stereoselective construction of ketosides^{3,4} as well as artificial ones⁵⁻⁸ has been enthusiastically investigated.

For the synthesis of ketosyl saccharides, several methods have been developed by our group.^{6,7} We have used *exo*-gly-cals in the construction of artificial sugar chains and reported the O-glycosidation of heptenitols,^{9–12} which are among the simplest *exo*-glycals formed by Tebbe-type methylenation¹³ of glyconolactone. The acid-promoted O-glycosidation of 2,6-anhydro-1-deoxyhept-1-enitols **1a–c** (Fig. 1) with



Figure 1. 2,6-Anhydro-1-deoxyhept-1-enitols as glycosyl donors.

methyl α -D-glucopyranosides **2** and **3** afforded 1-deoxyhept-2-ulosylglucosides, i.e., the 1'-C-methyl-substituted analogue of naturally occurring aldoside (Scheme 1). This glycosidation was promoted by various acids such as TiCl₄, SnCl₄, trimethylsilyl trifluoromethanesulfonate (TMSOTf), and trifluoromethanesulfonic acid (TfOH). The best result was achieved using TfOH as a promoter at -78 °C. As far as we have examined, the glycosidation of D-gluco-, D-galacto-, and D-manno-hept-1-enitols **1a-c** gave only α -ketoside, and the formation of β -ketoside was not detected. The glycosidation of peracetylated heptenitols with a 3-O-acetyl group also took place in an α -selective manner.



Scheme 1. Glycosidation of hept-1-enitols 1a-c.

Keywords: Ketoside; *exo*-Glycal; Heptenitol; Heptuloside; O-Glycosidation; Basic alumina.

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To investigate the generality of this α -selective glycosidation, we planned to examine the O-glycosidation of heptenitols using various glycosyl acceptors. We herein present a new library of hept-2-ulosylsaccharides and their X-ray crystallographic analyses.

2. Results and discussion

2.1. Glycosidation of heptenitols

As the ketosyl saccharides have quaternary anomeric carbons, we suspected that the unreactive axial hydroxyl group of the pyranose might not be glycosylated efficiently. We first examined the O-glycosidation of heptenitol **1a** with pyranosides **2**, **4**, **5**, and **6** (Table 1). The reaction was performed in CH₂Cl₂ at -78 °C in the presence of 10 mol % of TfOH. After quenching the reaction with triethylamine, the solvent was removed and the residue was subjected to column chromatography to afford the desired disaccharide in pure form. In the case of the primary alcohol **2**, the corresponding disaccharide was obtained in 97% yield with α -stereoselectivity.^{6a} The glucopyranoside **4**, which has an equatorial secondary hydroxyl group, showed similar reactivity (Table 1, entry 2). On the other hand, glycosidation with mannopyranosides **5** and **6**, which have axial secondary

Table 1. O-Glycosidation of 1a

E	BnO BnO BnO BnO BnO BnO	TfOH (10 mol %) MS4A CH ₂ Cl ₂ , –78 °C	BnO BnO BnO BnO BnO BnO
	donor (1.1 eq) acceptor 1a	(1.0 eq)	ŬŔ.
Entr	y Acceptor (ROH)	Time (min)	Yield (%)
1	BnO BnO 2 BnO 2 Me	60	97
2	BnO BnO BnO 4 HOOMe	50	94
3	BnO BnO 5 OMe	180	34
4	BnO OH BnO OMe	180	26
5	BnO BnO 7c	15	86

 $\begin{array}{c} BnO & \stackrel{6}{\longrightarrow} O \\ BnO & \stackrel{4}{\longrightarrow} O \\ BnO & \stackrel{4}{\longrightarrow} O \\ OBn \end{array} \qquad \begin{array}{c} BnO & \stackrel{O}{\longrightarrow} O \\ OBn \end{array}$

hydroxyl groups, proceeded very slowly (3 h) to give disaccharides in low yields (Table 1, entries 3 and 4). It is clear that pyranoside with an axially oriented hydroxyl group is reluctantly glycosylated. We next used the mannono-1,5lactone **7c** as an acceptor because we were interested in its conformational and electronic differentiation from ${}^{4}C_{1}$ pyranoside. Surprisingly, the reaction with **7c** proceeded rapidly (15 min), and the corresponding disaccharide was obtained in 86% yield with complete α -stereoselectivity. It should be noted that the secondary alcohol **7c** reacted more rapidly than the primary alcohol **2**. The rate of this glycosidation with glycosyl acceptors increased roughly in the following order:Axial secondary OH (**5**, **6**)≪equatorial secondary OH (**4**)≈primary OH (**2**)<secondary OH on lactone ring (**7c**).

Encouraged by the promising results obtained, we attempted the synthesis of heptulosylsaccharide using glycono-1,5lactones as the most reactive glycosyl acceptors. The use of lactones provides an additional advantage, enabling the transformation of the resulting disaccharides by methylenation,^{6b,13} reduction, or alkylation to afford a wide variety of sugar-related compounds.

Thus, the glycosidation of a combination of heptenitols 1a-c as glycosyl donors and sugar lactones $7-10^{14}$ derived from D-glucose, D-galactose, and D-mannose as acceptors (Fig. 2) was explored (Table 2).

Hydroxyl groups at C-2 in glyconolactones were smoothly glycosylated to afford heptulosyl- $(2 \rightarrow 2)$ -glyconolactones 11aa-cc in high yields (Table 2, entries 1-9). As we expected, C-2-OH in mannono-1,5-lactone reacted easily with the heptenitol 1a, 1b, and 1c giving the corresponding disaccharides in high yield (Table 2, entries 3, 6, 9). Similarly, C-3-OH and C-4-OH were efficiently glycosylated to afford the corresponding heptulosyl- $(2 \rightarrow 3)$ -glyconolactones 12 and heptulosyl- $(2 \rightarrow 4)$ -glyconolactones 13, respectively (Table 2, entries 10-27). Glycosidation with C-4-OH in galactono-1,5-lactone 9b also proceeded well and yielded the corresponding disaccharides (Table 2, entries 20, 23, 26). No remarkable differences in reactivity were observed in the glycosidation of gluco-, galacto-, and manno-heptenitols. In addition, it is noteworthy that all heptulosides were obtained as their respective single isomers.

The glycosidation of **1a–c** with primary alcohols **10a–c** was performed in the same manner (Table 2, entries 28–38). In entries 33 and 35 in Table 2, disaccharides were obtained in excellent yields within 15 min. In other cases, a longer reaction time was required for the completion of glycosidation. In addition, a longer reaction time caused the decomposition of heptenitols and heptulosides (Table 2, entries 30, 31, 37).



9b (Gal): 2-OBn(α), 4-OH(β)

9c (Man): 2-OBn(β), 4-OH(α)



10a (Glc): 2-OBn(α), 4-OBn(α) **10b** (Gal): 2-OBn(α), 4-OBn(β) **10c** (Man): 2-OBn(β), 4-OBn(α)

Figure 2. Glycono-1,5-lactones as glycosyl acceptors.

Table 2. O-Glycosidation of 1a-c with glycono-1,5-lactones

		1a-c donor (1.1 eq)	+	7a-c 8a-c 9a-c 10a-c acceptor (1.0 eq)	TfOH (10 mol%) MS4A CH ₂ Cl ₂ , –78 °C	11 12 13 14	
Donor	Acceptor	Time (min)			Product		Yield (%)
1a	7a	40				11aa Glc-(2→2)-Glc	89
1a	7b	30		BnO		11ab Glc- $(2 \rightarrow 2)$ -Gal	95
1a	7c	15		BnO	1'	11ac Glc- $(2 \rightarrow 2)$ -Man	86
1b	7a	15		BnO	<u>1</u> 2'	11ba Gal- $(2 \rightarrow 2)$ -Glc	99
1b	7b -	17		BnO-		11bb Gal- $(2 \rightarrow 2)$ -Gal	83
16	7c	17		BnOw	- Second	11bc Gal- $(2 \rightarrow 2)$ -Man	98
lc	7a	17			200	IIca Man- $(2 \rightarrow 2)$ -Glc	89
lc	76	15		11	2	11cb Man- $(2 \rightarrow 2)$ -Gal	95
Ic	7c	12				Hcc Man- $(2 \rightarrow 2)$ -Man	89
1a	8a	30				12aa Glc- $(2 \rightarrow 3)$ -Glc	93
1a	8b	15				12ab Glc- $(2 \rightarrow 3)$ -Gal	91
1a	8c	20		BnO	1'	12ac Glc- $(2 \rightarrow 3)$ -Man	81
1b	8a	20		BnO 2	OBn /	12ba Gal- $(2 \rightarrow 3)$ -Glc	88
1b	8b	15		BnO∣	$\langle 0 \rangle$	12bb Gal- $(2 \rightarrow 3)$ -Gal	89
1b	8c	30		BnO [≫]		12bc Gal- $(2 \rightarrow 3)$ -Man	77
1c	8a	15		40	3 OBn	12ca Man- $(2 \rightarrow 3)$ -Glc	84
1c	8b	15		12	OBI	12cb Man- $(2 \rightarrow 3)$ -Gal	88
1c	8c	30				12cc Man- $(2 \rightarrow 3)$ -Man	84
1a	9a	30				13aa Glc-(2→4)-Glc	99
1a	9b	30		BnO	1' 00	13ab Glc- $(2 \rightarrow 4)$ -Gal	81
1a	9c	20		BnO	' OBn	13ac Glc- $(2 \rightarrow 4)$ -Man	81
1b	9a	22		BnO	4 \ _0	13ba Gal- $(2 \rightarrow 4)$ -Glc	94
1b	9b	20		O [™] BnO		13bb Gal- $(2 \rightarrow 4)$ -Gal	82
1b	9c	20		ыю	OBn	13bc Gal- $(2 \rightarrow 4)$ -Man	83
lc	9a	16				13ca Man- $(2 \rightarrow 4)$ -Glc	97
1c	9b	15		13		13cb Man- $(2 \rightarrow 4)$ -Gal	97
1c	9c	30				13cc Man- $(2 \rightarrow 4)$ -Man	73
1a	10a	50				14aa Glc-(2→6)-Glc	72
1a	10a	10				14aa Glc- $(2 \rightarrow 6)$ -Glc	67
1a	10b	210		BnO ~ 0	, 1'	14ab Glc- $(2 \rightarrow 6)$ -Gal	77
1a	10c	120		BnO Dr	_	14ac Glc- $(2 \rightarrow 6)$ -Man	74
1a	10c	15		BnO /	7	14ac Glc- $(2 \rightarrow 6)$ -Man	82
1b	10a	14		BnO	1-0	14ba Gal- $(2 \rightarrow 6)$ -Glc	96
1b	10b	60		BnO-		14bb Gal- $(2 \rightarrow 6)$ -Gal	75

^a The amount of TfOH was increased to 15 mol %.

10c

10a

10b

10c

10

90

60

40

^b After 200 min, 5 mol % of TfOH was added.

1h

1c

1c

1c

35

36

37

38

In particular, glycosidation of gluco-heptenitol 1a with galactono-1,5-lactone 10b was quite slow, and therefore another portion of TfOH was added for the completion of the reaction (Table 2, entry 30). In entries 29 and 32 in Table 2, although the amount of TfOH was increased to 15 mol % to complete the reaction in a shorter time, yields were not increased.

It was surprising that the primary alcohols 10a-c were unreactive in some cases. The reactivity may depend on the combination of a glycosyl donor and an acceptor. For example, glycosidation of gluco-heptenitol 1a with mannonolactone 10c required 2 h (Table 2, entry 31), but otherwise the reaction of galacto-heptenitol 1b with 10c was completed in 10 min (Table 2, entry 35). Although the reactivity varied, heptulosyl- $(2 \rightarrow 6)$ -glyconolactones **14aa–cc** were obtained as their respective single isomers.

We therefore achieved the synthesis of 36 examples of 1-deoxy-2-heptulopyranosyl-glycono-1,5-lactone in a combination of heptenitols **1a–c** as glycosyl donors and **7–10** as acceptors. All gluco-, galacto-, and manno-heptuloside products were isolated as their respective single anomeric isomers.

14bc Gal- $(2 \rightarrow 6)$ -Man

14ca Man- $(2 \rightarrow 6)$ -Glc

14cb Man- $(2 \rightarrow 6)$ -Gal

14cc Man- $(2 \rightarrow 6)$ -Man

OBn

14

97

69

70

79

2.2. Determination of the configurations at anomeric position

The configurations at C-2' anomeric position of hept-2ulosides were determined by X-ray crystallographic analyses and NMR study.

manno-Hept-2-uloside 14cc was obtained as colorless crystals. The X-ray crystallographic structure of 14cc is shown in Figure 3.15 Based on this X-ray crystallographic analysis, it was elucidated that manno-heptuloside possessed a 5C2 chair conformation and α -anomeric configuration. Additionally, the mannonolactone part had a B_{2.5} boat conformation. As only one isomer was formed in every case, it is suggested that other *manno*-heptulosides also possess the α -anomeric configuration.



Figure 3. X-ray crystal structure of 14cc. Hydrogen atoms are omitted for clarity.

The anomeric configuration of *gluco*- and *galacto*- heptulopyranosides was anticipated by analyzing the vicinal C–H coupling constants ${}^{3}J_{C,H}$ in NMR spectra.^{10,16} It is known that the magnitude of ${}^{3}J_{C,H}$ depends on the dihedral angle between the C-1/C-2 bond and C-3/H-3 bond. As shown in Figure 4, the C-1' exocyclic carbon and H-3' axial proton

β-anome

 $RO \xrightarrow{HO} 1' CH_3 \qquad I$ $RO \xrightarrow{2'} CH_3 \qquad I$ $\alpha \text{-anomer}$ **11aa** Glc-(2 \rightarrow 2)-Glc 1.9 Hz **12ba** Gal-(2 \rightarrow 3)-Glc 2.0 Hz **13ab** Glc-(2 \rightarrow 4)-Gal 1.8 Hz

19 Hz

Figure 4. Vicinal coupling constants ${}^{3}J_{C-1',H-3'}$.

Table 3. Deprotection of 11cc and 12cc

14bb Gal-(2→6)-Gal

are oriented synclinally in α -ketoside and antiperiplanarly in β -ketoside. By the measurement of the vicinal coupling constant ${}^{3}J_{C-1',H-3'}$ of our products, which were approximately 1.8–2.0 Hz, it is obvious that newly formed glycosidic bonds have the α -configuration. We can be fairly certain that the O-glycosidation of *gluco-*, *galacto-*, and *manno-*heptenitols gave α -glycosides.

2.3. Deprotection

We then focused on the deprotection of protected hydroxyl groups. In preliminary experiments, it was found that the removal of benzyl groups by hydrogenolysis was accompanied by the cleavage of ketoside bonds. Other reactions such as Birch reduction, catalytic hydrogen transfer,¹⁷ the use of lithium naphthalenide,¹⁸ and oxidation^{19–21} gave unsatisfactory results.

We therefore returned to the use of Pd-catalyzed hydrogenolysis. The deprotection of Man- $(2 \rightarrow 3)$ -Man 12cc was examined in various solvents using Pd/C or Pd(OH)₂/C as a catalyst (Table 3). When the reaction proceeded in MeOH, solvolysis occurred to afford 17 (Table 3, entry 1). Using less nucleophilic trifluoroethanol (TFE) as a solvent, intramolecular cyclization followed by cleavage of the ketoside bond gave the cyclized product 18 quantitatively (Table 3, entry 2). It was speculated that the acidity of TFE caused the cleavage of ketoside. Similarly, the use of THF as a solvent resulted in the formation of 18 (Table 3, entry 3). On the hypothesis that the formation of a hydrogen bond would prevent intramolecular nucleophilic attack of the anomeric center, hydrogenolysis was performed in mixed solvents THF/H₂O, and the desired product 15cc was obtained in 44% yield (Table 3, entry 4).

Although the use of such mixed solvents appeared promising, unexpected solvolysis occurred in the case of Man- $(2 \rightarrow 2)$ -Man **11cc** (Table 3, entry 6). It was found that deprotected disaccharides vary in stability. In order to prevent solvolysis, slightly basic conditions were examined. In the presence of



Entry	Substrate	Pd	Cat. (wt %)	Solvent	Time (h)	Yield (%)	Other products
1	12cc	Pd/C	20	MeOH	4	0	17
2		Pd(OH) ₂ /C	25	TFE	48	0	18
3		Pd/C	50	THF	9	0	18
4		Pd(OH) ₂ /C	20	THF/H ₂ O	15	44	
5 ^a		Pd(OH) ₂ /C	50	THF; MeOH	70	97	
6	11cc	Pd(OH) ₂ /C	50	THF/H ₂ O	2	0	19
7 ^a		Pd(OH) ₂ /C	50	THF; MeOH	37	65	

^a Hydrogenation was performed in the presence of basic alumina (50 wt %).

organic or inorganic bases such as pyridine, triethylamine, sodium acetate, and potassium carbonate, hydrogenolysis did not proceed smoothly.²² We finally found that basic alumina was effective: it prevented undesired side reactions and maintained the catalyst activity. In the presence of basic alumina, the solubility of substrates in solvent markedly affected the reaction rate. In the early stage of hydrogenolysis, the reaction proceeds much faster in THF than in MeOH or *t*-BuOH. Then, as debenzylation progresses, the solubility in THF becomes low, requiring MeOH to be added. Under these conditions, both **12cc** and **11cc** were debenzylated in satisfactory yields (Table 3, entries 5 and 7).

Thus, debenzylation of 36 heptulosides **11–14** was explored using catalytic hydrogenation in the presence of Pd(OH)₂/C and basic alumina (Table 4). In many cases, debenzylation afforded the desired deprotected disaccharides. However, some substrates were quite unstable even under basic conditions, and deprotected disaccharides could not be isolated. Particularly, the removal of benzyl groups in heptulosyl- $(2 \rightarrow 6)$ -glyconolactone was difficult. It appears that deprotected heptulosides **21** are fairly unstable.²³ Although considerable effort was made, further improvement was not achieved in these cases except for Gal- $(2 \rightarrow 6)$ -Gal **14b**. Ultimately, the 25 disaccharides shown in Table 4 were obtained in a deprotected form.

3. Conclusion

We achieved the synthesis of disaccharides, hept-2-ulosylglyconolactones, by acid-promoted O-glycosidation of 2,6anhydro-1-deoxyhept-1-enitols **1a–c**. Under the reaction conditions described here, O-glycosidation of heptenitols **1a–c** proceeded with complete stereoselectivity, and each disaccharide was isolated as a single anomeric isomer. Their configurations at anomeric position were identified as α -stereochemistry by X-ray crystallography and NMR vicinal coupling constant ${}^{3}J_{C-1',H-3'}$. Although the removal of benzyl groups was problematic, we found that Pd-catalyzed hydrogenolysis in the presence of basic alumina was effective in preventing cleavage of the ketoside bonds.

4. Experimental

4.1. General methods

Melting points were measured on a YANACO Micro Melting Point Apparatus and were uncorrected. IR spectra were recorded on a Jasco FT-IR-8000 Fourier-transform infrared spectrometer. ¹H and ¹³C NMR, and three-bond carbon-proton coupling constants ³ $J_{C,H}$ were measured on a JEOL ECP 600 (600 MHz) NMR spectrometer in CDCl₃, CD₃OD,

Table 4. Deprotection

	BnO BnO BnO BnO BnO	$\begin{array}{c} \begin{array}{c} Pd(OH)_2/C, \text{ basic alumina} \\ H_2 \\ \hline \\ O \end{array} \end{array} $	HO HO HO O	
Entry	O-benzylated disaccharide	Product		Yield (%)
1 2 3 4 5 6 7 8 9	11aa Glc- $(2 \rightarrow 2)$ -Glc 11ab Glc- $(2 \rightarrow 2)$ -Gal 11ac Glc- $(2 \rightarrow 2)$ -Man 11ba Gal- $(2 \rightarrow 2)$ -Glc 11bb Gal- $(2 \rightarrow 2)$ -Gal 11ca Man- $(2 \rightarrow 2)$ -Glc 11cb Man- $(2 \rightarrow 2)$ -Gal 11cc Man- $(2 \rightarrow 2)$ -Man		16aa Glc- $(2 \rightarrow 2)$ -Glc 16ab Glc- $(2 \rightarrow 2)$ -Gal 16ac Glc- $(2 \rightarrow 2)$ -Man 16ba Gal- $(2 \rightarrow 2)$ -Glc 16bb Gal- $(2 \rightarrow 2)$ -Gal 16bc Gal- $(2 \rightarrow 2)$ -Gal 16bc Man- $(2 \rightarrow 2)$ -Glc 16cb Man- $(2 \rightarrow 2)$ -Glc 16cb Man- $(2 \rightarrow 2)$ -Gal 16cc Man- $(2 \rightarrow 2)$ -Gal 16cc Man- $(2 \rightarrow 2)$ -Man	quant. 89 quant. 78 quant. quant. 89 83 65
10 11 12 13 14 15 16	12aa Glc- $(2 \rightarrow 3)$ -Glc 12ab Glc- $(2 \rightarrow 3)$ -Gal 12ba Gal- $(2 \rightarrow 3)$ -Glc 12bb Gal- $(2 \rightarrow 3)$ -Glc 12cb Man- $(2 \rightarrow 3)$ -Glc 12cb Man- $(2 \rightarrow 3)$ -Gal 12cc Man- $(2 \rightarrow 3)$ -Man	HO HO HO HO HO HO HO HO HO HO HO HO HO H	15aa Glc- $(2 \rightarrow 3)$ -Glc 15ab Glc- $(2 \rightarrow 3)$ -Gal 15ba Gal- $(2 \rightarrow 3)$ -Glc 15bb Gal- $(2 \rightarrow 3)$ -Gal 15ca Man- $(2 \rightarrow 3)$ -Glc 15cb Man- $(2 \rightarrow 3)$ -Gal 15cc Man- $(2 \rightarrow 3)$ -Man	quant. 99 97 quant. 98 83 65
17 18 19 20 21 22 23 24	13aa Glc- $(2 \rightarrow 4)$ -Glc 13ab Glc- $(2 \rightarrow 4)$ -Gal 13ac Glc- $(2 \rightarrow 4)$ -Man 13ba Gal- $(2 \rightarrow 4)$ -Glc 13bb Gal- $(2 \rightarrow 4)$ -Gal 13bc Gal- $(2 \rightarrow 4)$ -Gal 13cb Man- $(2 \rightarrow 4)$ -Gal 13cc Man- $(2 \rightarrow 4)$ -Man	HO HO HO HO HO HO HO HO HO HO HO HO HO H	20aa Glc- $(2 \rightarrow 4)$ -Glc 20ab Glc- $(2 \rightarrow 4)$ -Gal 20ac Glc- $(2 \rightarrow 4)$ -Man 20ba Gal- $(2 \rightarrow 4)$ -Glc 20bb Gal- $(2 \rightarrow 4)$ -Gal 20bc Gal- $(2 \rightarrow 4)$ -Gal 20cb Man- $(2 \rightarrow 4)$ -Gal 20cc Man- $(2 \rightarrow 4)$ -Man	quant. 91 quant. 75 quant. quant. 94 quant.
25	14bb Gal-(2→6)-Gal		21bb Gal-(2→6)-Gal	59

or D₂O solutions. Mass spectra (MS) and high-resolution mass spectra (HRMS) were recorded with a JEOL JMS-SX102A mass spectrometer with FAB using 3-nitrobenzyl alcohol (NBA) as the matrix. Optical rotations were measured with a Jasco DIP-370 digital polarimeter. X-ray crystallographic measurements were performed with a Rigaku RAXIS-RAPID Imaging Plate Diffractometer with graphite monochromated MoK_{α} radiation at 123 K. TLC was performed on precoated plates (Merck TLC Aluminum sheets silica 60 F₂₅₄) with detection by UV light or with phosphomolybdic acid in EtOH/H₂O followed by heating. Column chromatography was performed using SiO₂ (Silica Gel 60 N, spherical, neutral, Kanto).

4.2. General procedure 1: glycosidation of heptenitols with alcohols

To a stirred mixture of hept-1-enitol **1a** (237.0 mg, 0.44 mmol), hydroxylactone **7a** (180.0 mg, 0.40 mmol), and molecular sieves 4 Å (400.0 mg) in CH₂Cl₂ (8.0 mL) was added TfOH (3.5 μ L, 0.04 mmol) at -78 °C. The reaction mixture was stirred at -78 °C and then quenched with triethylamine. After the removal of the solvent, the residue was purified by column chromatography (silica gel, neutral, hexane/ethyl acetate 3:1) to give disaccharide **11aa** (350.0 mg).²⁴

4.2.1. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-gluco-hept-2ulosyl-(2→2)-3,4,6-tri-*O*-benzyl-D-glucono-1,5-lactone (**11aa**). Colorless syrup; $[α]_D^{26}$ +64.9 (*c* 1.16, CHCl₃); IR (neat): 3031, 2913, 2869, 1763, 1105, 1082, 1028, 1029, 739, 700 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4096.

4.2.2. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-*gluco*-hept-2ulosyl-($2 \rightarrow 2$)-3,4,6-tri-*O*-benzyl-D-galactono-1,5-lactone (**11ab**). Colorless needle; mp 81–83 °C; [α]_D²⁵ +70.6 (*c* 0.5, CHCl₃); IR (KBr): 3031, 2913, 2869, 1761, 1497, 1455, 1362, 735, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4096.

4.2.3. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-gluco-hept-2ulosyl-($2 \rightarrow 2$)-3,4,6-tri-*O*-benzyl-D-mannono-1,5-lactone (**11ac**). Colorless syrup; $[\alpha]_D^{27}$ +47.6 (*c* 1.15, CHCl₃); IR (neat): 3031, 2928, 2867, 1775, 1455, 1127, 1082, 1065, 1028, 737, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4076.

4.2.4. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy- α -D-*galacto*-hept-**2-ulosyl-(2** \rightarrow **2)-3,4,6-tri-***O***-benzyl-D-glucono-1,5-lactone (11ba).** Colorless syrup; $[\alpha]_D^{29}$ +59.4 (*c* 1.21, CHCl₃); IR (neat): 3031, 2917, 2869, 1763, 1455, 1100, 737, 700 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4096.

4.2.5. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-galacto-hept-2-ulosyl-(2→2)-3,4,6-tri-*O*-benzyl-D-galactono-1,5-lactone (11bb). Colorless syrup; $[α]_D^{26}$ +52.5 (*c* 1.05, CHCl₃); IR (neat): 3031, 2923, 2874, 1763, 1455, 1101, 739, 700 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4084.

4.2.6. 3,4,5,7-Tetra-O-benzyl-1-deoxy- α -D-galacto-hept-2-ulosyl-(2 \rightarrow 2)-3,4,6-tri-O-benzyl-D-mannono-1,5-lactone (11bc). Colorless syrup; $[\alpha]_D^{27}$ +48.9 (c 1.03, CHCl₃);

IR (neat): 3031, 2915, 2867, 1775, 1455, 1096, 1059, 737, 698 cm⁻¹; HRMS (FAB): calcd for $C_{62}H_{64}O_{11}K$ 1023.4086, found 1023.4061.

4.2.7. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-*manno*-hept-2-ulosyl-(2 \rightarrow 2)-3,4,6-tri-*O*-benzyl-D-glucono-1,5-lactone (11ca). Colorless syrup; $[\alpha]_D^{27}$ +49.1 (*c* 1.06, CHCl₃); IR (neat): 3031, 2911, 2869, 1763, 1455, 1098, 1078, 737, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4086.

4.2.8. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-*manno*-hept-2-ulosyl-($2 \rightarrow 2$)-3,4,6-tri-*O*-benzyl-D-galactono-1,5-lactone (11cb). Colorless syrup; $[\alpha]_D^{26}$ +57.7 (*c* 1.13, CHCl₃); IR (neat): 3031, 2915, 2867, 1758, 1455, 1096, 737, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4105.

4.2.9. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-*manno*-hept-2-ulosyl-($2 \rightarrow 2$)-3,4,6-tri-*O*-benzyl-D-mannono-1,5-lactone (11cc). Colorless syrup; $[\alpha]_D^{25}$ +65.5 (*c* 1.01, CHCl₃); IR (neat): 3031, 2923, 2867, 1777, 1455, 1073, 739, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4084.

4.2.10. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-gluco-hept-2ulosyl-(2 \rightarrow 3)-2,4,6-tri-*O*-benzyl-D-glucono-1,5-lactone (12aa). Colorless syrup; $[\alpha]_D^{26}$ +98.8 (*c* 1.14, CHCl₃); IR (neat): 3031, 2927, 2863, 1754, 1455, 1073, 737, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁Na 1007.4346, found 1007.4355.

4.2.11. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-*gluco*-hept-2ulosyl-(2→3)-2,4,6-tri-*O*-benzyl-D-galactono-1,5-lactone (12ab). Colorless syrup; $[α]_D^{30}$ +90.9 (*c* 1.04, CHCl₃); IR (neat): 3032, 2921, 2867, 1750, 1455, 1096, 737, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4075.

4.2.12. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-gluco-hept-2ulosyl-($2 \rightarrow 3$)-2,4,6-tri-*O*-benzyl-D-mannono-1,5-lactone (**12ac**). Colorless syrup; $[\alpha]_D^{27}$ +35.3 (*c* 1.08, CHCl₃); IR (neat): 3031, 2924, 2869, 1765, 1455, 1125, 1088, 739, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4085.

4.2.13. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy- α -D-galacto-hept-**2-ulosyl-(** $2 \rightarrow 3$ **)-2,4,6-tri-***O*-benzyl-D-glucono-1,5-lactone (**12ba**). Colorless syrup; $[\alpha]_D^{26}$ +101.7 (*c* 1.08, CHCl₃); IR (neat): 3031, 2924, 2870, 1754, 1455, 1096, 1069, 737, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁Na 1007.4346, found 1007.4354.

4.2.14. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy- α -D-galacto-hept-2-ulosyl-($2 \rightarrow 3$)-2,4,6-tri-*O*-benzyl-D-galactono-1,5-lactone (12bb). Colorless needle; mp 127–128 °C; $[\alpha]_{D}^{24}$ +101.0 (*c* 1.0, CHCl₃); IR (KBr): 3031, 2867, 1732, 1497, 1098, 735, 696 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4094.

4.2.15. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy- α -D-galacto-hept-2-ulosyl-(2 \rightarrow 3)-2,4,6-tri-*O*-benzyl-D-mannono-1,5-lactone (12bc). Colorless syrup; $[\alpha]_D^{28}$ +44.2 (*c* 1.28, CHCl₃); IR (neat): 3031, 2915, 2869, 1765, 1455, 1098, 737, 698 cm⁻¹; HRMS (FAB): calcd for $C_{62}H_{64}O_{11}K$ 1023.4086, found 1023.4082.

4.2.16. 3,4,5,7-Tetra-*O*-**benzyl**-1-**deoxy**-α-D-*manno*-**hept**-**2-ulosyl**-($2 \rightarrow 3$)-**2,4,6-tri**-*O*-**benzyl**-D-**glucono**-1,**5**-lactone (**12ca**). Colorless syrup; $[\alpha]_D^{27}$ +68.3 (*c* 1.18, CHCl₃); IR (neat): 3031, 2911, 2867, 1757, 1455, 1092, 1073, 745, 702 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4093.

4.2.17. 3,4,5,7-Tetra-*O*-**benzyl**-1-deoxy-α-D-*manno*-hept-**2-ulosyl**-(2→3)-2,4,6-tri-*O*-benzyl-D-galactono-1,5-lactone (12cb). Colorless syrup; $[\alpha]_D^{26}$ +77.9 (*c* 1.10, CHCl₃); IR (neat): 3031, 2919, 2869, 1750, 1455, 1119, 1071, 752, 735, 696 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4064.

4.2.18. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-manno-hept-2-ulosyl-($2 \rightarrow 3$)-2,4,6-tri-*O*-benzyl-D-mannono-1,5-lactone (12cc). Colorless syrup; $[\alpha]_D^{30}$ +39.7 (*c* 1.16, CHCl₃); IR (neat): 3031, 2915, 2867, 1773, 1455, 1113, 739, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4083.

4.2.19. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy- α -D-gluco-hept-2ulosyl-(2 \rightarrow 4)-2,3,6-tri-*O*-benzyl-D-glucono-1,5-lactone (13aa). Colorless syrup; $[\alpha]_D^{28}$ +47.6 (*c* 1.09, CHCl₃); IR (neat): 3031, 2924, 2869, 1792, 1455, 1088, 735, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1007.4346, found 1007.4331.

4.2.20. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-*gluco*-hept-2ulosyl-(2→4)-2,3,6-tri-*O*-benzyl-D-galactono-1,5-lactone (13ab). Colorless syrup; $[\alpha]_D^{27}$ +29.8 (*c* 1.13, CHCl₃); IR (neat): 3031, 2924, 2867, 1791, 1455, 1090, 7367, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4074.

4.2.21. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy- α -D-gluco-hept-2ulosyl-(2 \rightarrow 4)-2,3,6-tri-*O*-benzyl-D-mannono-1,5-lactone (13ac). Colorless syrup; $[\alpha]_D^{27}$ +26.8 (*c* 1.06, CHCl₃); IR (neat): 3031, 2924, 2869, 1769, 750, 739, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4111.

4.2.22. 3,4,5,7-Tetra-*O*-**benzyl-1**-**deoxy**-α-D-*galacto*-**hept-2**-**ulosyl**-(2 → 4)-2,3,6-tri-*O*-**benzyl**-D-**glucono**-1,5-**lactone** (**13ba**). Colorless syrup; $[\alpha]_D^{30}$ +60.6 (*c* 0.83, CHCl₃); IR (neat): 3031, 2921, 2870, 1790, 1455, 1098, 737, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4082.

4.2.23. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-galacto-hept-**2-ulosyl-(2→4)-2,3,6-tri**-*O*-benzyl-D-galactono-1,5-lactone (13bb). Colorless syrup; $[\alpha]_D^{30}$ +27.4 (*c* 0.85, CHCl₃); IR (neat): 3031, 2923, 2870, 1788, 1455, 1100, 737, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4083.

4.2.24. 3,**4**,**5**,**7**-**Tetra**-*O*-**benzyI**-**1**-**deoxy**-*α*-**D**-*galacto*-**hept**-**2**-**ulosyI**-(2 → 4)-2,**3**,**6**-tri-*O*-**benzyI**-**D**-**mannono-1,5**-**lactone** (**13bc**). Colorless syrup; $[\alpha]_D^{27}$ +28.6 (*c* 1.02, CHCl₃); IR (neat): 3031, 2923, 2872, 1769, 1455, 1101, 1082, 735, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4081.

4.2.25. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-*manno*-hept-**2-ulosyl-(2**→4)-2,3,6-tri-*O*-benzyl-D-glucono-1,5-lactone (13ca). Colorless syrup; $[\alpha]_D^{27}$ +36.9 (*c* 1.06, CHCl₃); IR (neat): 3031, 2915, 2867, 1786, 1455, 1100, 1084, 745, 700 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4087.

4.2.26. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-manno-hept-2-ulosyl-($2 \rightarrow 4$)-2,3,6-tri-*O*-benzyl-D-galactono-1,5-lactone (13cb). Colorless syrup; $[\alpha]_D^{27}$ –0.43 (*c* 1.07, CHCl₃); IR (neat): 3031, 2907, 2869, 1788, 1455, 1111, 739, 700 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4074.

4.2.27. 3,**4**,**5**,**7**-**Tetra**-*O*-**benzyl**-**1**-**deoxy**-α-**D**-*manno*-**hept**-**2**-**ulosyl**-(**2**→**4**)-**2**,**3**,**6**-tri-*O*-**benzyl**-**D**-**mannono**-**1**,**5**-lactone (13cc). Colorless syrup; $[\alpha]_D^{30}$ +9.8 (*c* 1.05, CHCl₃); IR (neat): 3031, 2924, 2870, 1767, 1455, 1109, 748, 737, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4080.

4.2.28. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-D-*gluco*-hept-2ulosyl-($2 \rightarrow 6$)-2,3,4-tri-*O*-benzyl-D-glucono-1,5-lactone (14aa). Colorless syrup; mp 79–81 °C; $[\alpha]_D^{25}$ +70.3 (*c* 1.00, CHCl₃); IR (KBr): 3031, 2867, 1759, 1455, 1123, 1071, 739, 696 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4102.

4.2.29. 3,4,5,7-Tetra-*O*-**benzyl-1**-**deoxy**-**D**-*gluco*-**hept-2**-**ulosyl-(2** \rightarrow **6)-2,3,4-tri**-*O*-**benzyl**-**D**-**galactono-1,5-lactone** (**14ab**). Colorless syrup; $[\alpha]_D^{25}$ +78.8 (*c* 1.17, CHCl₃); IR (neat): 3031, 2921, 2869, 1752, 1455, 1067, 735, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁Na 1007.4346, found 1007.4333.

4.2.30. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-D-*gluco*-hept-2ulosyl-($2 \rightarrow 6$)-2,3,4-tri-*O*-benzyl-D-mannono-1,5-lactone (14ac). Colorless syrup; $[\alpha]_D^{29}$ +19.8 (*c* 1.07, CHCl₃); IR (neat): 3031, 2928, 2870, 1775, 1455, 1096, 739, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4088, found 1023.4066.

4.2.31. 3,4,5,7-Tetra-*O***-benzyl-1-deoxy-D***-galacto***-hept-2-ulosyl-(2\rightarrow6)-2,3,4-tri-***O***-benzyl-D**-glucono-1,5-lactone (**14ba**). Colorless syrup; $[\alpha]_D^{27}$ +66.6 (*c* 1.12, CHCl₃); IR (neat): 3031, 2919, 2874, 1757, 1455, 1096, 739, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4086.

4.2.32. 3,4,5,7-Tetra-*O***-benzyl-1-deoxy-α-D***-galacto***-hept-2-ulosyl-(2→6)-2,3,4-tri***-O***-benzyl-D**-galactono-1,5-lactone (14bb). Colorless syrup; $[\alpha]_D^{27}$ +75.8 (*c* 1.19, CHCl₃); IR (neat): 3031, 2917, 2869, 1750, 1455, 1109, 1057, 745, 737, 745, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁Na 1007.4346, found 1007.4362.

4.2.33. 3,4,5,7-Tetra-*O***-benzyl-1-deoxy-***α***-D***-galacto***-hept-2-ulosyl-(2→6)-2,3,4-tri-***O***-benzyl-D-mannono-1,5-lactone (14bc).** Colorless syrup; $[\alpha]_D^{26}$ +23.1 (*c* 1.14, CHCl₃); IR (neat): 3031, 2917, 2870, 1775, 1455, 1113, 1098, 735, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4099.

4.2.34. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy- α -D-manno-hept-2-ulosyl-($2 \rightarrow 6$)-2,3,4-tri-*O*-benzyl-D-glucono-1,5-lactone (14ca). Colorless syrup; $[\alpha]_D^{27}$ +57.2 (*c* 1.32, CHCl₃); IR

(neat): 3031, 2921, 2867, 1757, 1455, 1094, 1078, 735, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4081.

4.2.35. 3,4,5,7-Tetra-*O*-benzyl-1-deoxy-α-D-manno-hept-2-ulosyl-(2→6)-2,3,4-tri-*O*-benzyl-D-galactono-1,5-lactone (14cb). Colorless syrup; $[\alpha]_D^{27}$ +56.8 (*c* 1.28, CHCl₃); IR (neat): 3031, 2921, 2869, 1750, 1455, 1104, 737, 698 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4101.

4.2.36. 3,**4**,**5**,**7**-**Tetra**-*O*-**benzyl**-**1**-**deoxy**-*α*-**D**-*manno*-**hept**-**2**-**ulosyl**-(**2**→**6**)-**2**,**3**,**4**-**tri**-*O*-**benzyl**-**D**-**mannono**-**1**,**5**-**lac**-**tone** (**14cc**). Colorless block; mp 111 °C; $[α]_D^{27}$ +22.7 (*c* 0.92, CHCl₃); IR (KBr): 3032, 2936, 2346, 1775, 1499, 1456, 1109, 739, 700 cm⁻¹; HRMS (FAB): calcd for C₆₂H₆₄O₁₁K 1023.4086, found 1023.4168.

4.3. General procedure 2: deprotection

To a solution of **11aa** (73.0 mg, 0.07 mmol) in THF (2.5 mL) were added basic alumina (18.0 mg) and 20% Pd(OH)₂/C (37.0 mg) under an argon atmosphere, and the mixture was stirred under a hydrogen atmosphere (balloon) at room temperature. After 20 h, to the mixture was added MeOH (0.3 mL) and stirred for 28 h. The reaction mixture was filtered through filter paper, and the filtrate was evaporated and dried to give **16aa** (28.0 mg).

4.3.1. 1-Deoxy-D-*gluco*-hept-2-ulosyl- $(2 \rightarrow 2)$ -D-glucono-**1,5-lactone** (**16aa**). Colorless amorphous solid; $[\alpha]_D^{23}$ +96.2 (*c* 1.24, CH₃OH); ¹H NMR (600 MHz, CD₃OD): δ 4.61 (m, 1H), 4.52 (m, 1H), 4.48 (m, 1H), 3.99 (m, 1H), 3.86–3.58 (m, 8H), 1.53 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): δ 176.0, 103.5, 81.5, 77.8, 75.2, 75.1, 74.0, 73.8, 71.8, 71.4, 64.4, 62.8, 23.2.

4.3.2. 1-Deoxy-\alpha-D-*gluco*-hept-2-ulosyl-($2 \rightarrow 2$)-D-galactono-1,5-lactone (16ab). Colorless amorphous solid; $[\alpha]_D^{27}$ +60.8 (*c* 0.23, CH₃OH); ¹H NMR (600 MHz, CD₃OD): δ 4.67 (d, *J*=8.8 Hz, 1H, H-2), 4.36 (dd, *J*=8.3, 8.8 Hz, 1H), 4.15 (dd, *J*=2.2, 8.8 Hz, 1H, H-3), 3.82 (ddd, *J*=2.2, 5.8, 10.2 Hz, 1H), 3.74 (dd, *J*=2.5, 11.6 Hz, 1H), 3.70 (m, 1H), 3.58–3.52 (m, 4H), 3.23 (m, 1H), 3.09 (d, *J*=9.9 Hz, 1H, H-3'), 1.47 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): δ 175.0, 103.3, 81.2, 78.1, 75.7, 75.1, 74.4, 73.4, 71.7, 70.3, 63.5, 62.5, 22.8.

4.3.3. 1-Deoxy- α -**D**-gluco-hept-2-ulosyl-($2 \rightarrow 2$)-**D**-mannono-1,5-lactone (16ac). Colorless amorphous solid; [α]_D²⁴ +118.6 (*c* 0.5, CH₃OH); ¹H NMR (600 MHz, CD₃OD): δ 4.86 (d, *J*=2.8 Hz, 1H, H-2), 4.12 (ddd, *J*=8.3, 5.8, 2.5 Hz, 1H, H-5), 3.95 (dd, *J*=1.1, 2.8 Hz, 1H, H-3), 3.87 (ddd, *J*=9.9, 2.2, 6.6 Hz, 1H, H-6'), 3.78 (dd, *J*=9.1, 9.6 Hz, 1H, H-4'), 3.75 (dd, *J*=2.5, 12.4 Hz, 1H, H-6), 3.71 (dd, *J*=2.2, 11.8 Hz, 1H, H-7'), 3.7 (dd, *J*=1.1, 8.3 Hz, 1H, H-4), 3.64 (dd, *J*=6.6, 11.8 Hz, 1H, H-7'), 3.49 (dd, *J*=5.8, 12.4 Hz, 1H, H-6), 3.13 (dd, *J*=9.9, 9.1 Hz, 1H, H-5'), 3.11 (d, *J*=9.6 Hz, 1H, H-3'), 1.45 (s, 3H, -CH₃); ¹³C NMR (150 MHz, CD₃OD): δ 172.5, 103.1, 82.2, 78.3, 77.2, 75.3, 74.9, 72.1, 70.8, 69.9, 63.0, 62.8, 22.7.

4.3.4. 1-Deoxy- α -D-galacto-hept-2-ulosyl-($2 \rightarrow 2$)-D-glucono-1,5-lactone (16ba). Colorless amorphous solid; ¹H NMR (600 MHz, CD₃OD): δ 4.52–4.47 (m, 2H), 4.39 (dd, J=4.7, 5.0 Hz, 1H), 3.91 (m, 1H), 3.80 (m, 2H), 3.70–3.66 (m, 2H), 3.63–3.55 (m, 3H), 3.49 (d, J=10.2 Hz, 1H, H-3'), 1.46 (s, 3H, H-1').

4.3.5. 1-Deoxy-\alpha-D-*galacto*-hept-2-ulosyl-(2 \rightarrow 2)-D-galactono-1,5-lactone (16bb). Colorless amorphous solid; $[\alpha]_{28}^{28}$ +60.2 (*c* 1.01, CH₃OH); ¹H NMR (600 MHz, CD₃OD): δ 4.68 (d, *J*=8.8 Hz, 1H, H-2), 4.35 (dd, *J*=8.5, 8.5 Hz, 1H), 4.16 (dd, *J*=2.2, 8.8 Hz, 1H, H-3), 4.04 (m, 1H), 3.81 (m, 1H), 3.71–3.65 (m, 3H), 3.61–3.51 (m, 4H), 1.49 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): δ 175.0, 103.6, 81.2, 75.7, 74.7, 73.4, 73.0, 71.7, 71.2, 70.4, 63.5, 62.9, 22.7.

4.3.6. 1-Deoxy-α-D-galacto-hept-2-ulosyl-($2 \rightarrow 2$)-D-mannono-1,5-lactone (16bc). Colorless amorphous solid; ¹H NMR (600 MHz, CD₃OD): δ 4.91 (d, J=2.5 Hz, 1H, H-2), 4.16 (ddd, J=8.3, 2.5, 5.8 Hz, 1H, H-5), 4.12 (ddd, J=1.4, 4.1, 7.7 Hz, 1H, H-6'), 4.00 (dd, J=2.5, 0.8 Hz, 1H, H-3), 3.95 (dd, J=9.9, 3.3 Hz, 1H, H-4'), 3.85 (dd, J=3.3, 1.4 Hz, 1H, H-5'), 3.79 (dd, J=2.5, 12.4 Hz, 1H, H-6), 3.75 (dd, J=0.8, 8.3 Hz, 1H, H-4), 3.69 (dd, J=5.8, 12.4 Hz, 1H, H-6), 3.67 (dd, J=7.7, 11.5 Hz, 1H, H-7'), 3.57 (d, J=9.9 Hz, 1H, H-3'), 3.56 (dd, J=4.1, 11.5 Hz, 1H, H-7'), 1.45 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): δ 172.8, 103.4, 82.2, 77.3, 74.8, 74.2, 71.6, 71.6, 70.8, 69.8, 63.3, 62.8, 22.6.

4.3.7. 1-Deoxy-\alpha-D-manno-hept-2-ulosyl-(2 \rightarrow 2)-D-glucono-1,5-lactone (16ca). Colorless amorphous solid; ¹H NMR (600 MHz, CD₃OD): δ 4.46 (m, 1H), 4.38 (m, 1H), 3.90 (m, 1H), 3.77 (dd, *J*=1.7, 11.8 Hz, 1H), 3.72 (dd, *J*=3.3, 9.1 Hz, 1H, H-4'), 3.68 (dd, *J*=3.9, 11.8 Hz, 1H), 3.63–3.59 (m, 3H), 3.55 (d, *J*=3.3 Hz, 1H, H-3'), 3.51 (dd, *J*=1.9, 6.1 Hz, 1H), 3.47 (dd, *J*=9.1, 9.4 Hz, 1H, H-5'), 1.41 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): δ 176.0, 104.6, 81.4, 76.0, 74.4, 73.8, 73.7, 72.5, 71.2, 67.9, 64.3, 63.0, 22.3.

4.3.8. 1-Deoxy-\alpha-*D***-***manno***-hept-2-ulosyl-(2 \rightarrow 2)-***D***-galactono-1,5-lactone (16cb). Colorless amorphous solid; ¹H NMR (600 MHz, CD₃OD): \delta 4.69 (d,** *J***=8.8 Hz, 1H, H-2), 4.32 (dd,** *J***=8.5, 8.5 Hz, 1H, H-6), 4.14 (dd,** *J***=2.5, 8.8 Hz, 1H, H-3), 3.82 (ddd,** *J***=2.2, 5.8, 9.9 Hz, 1H, H-6'), 3.78 (dd,** *J***=3.3, 9.6 Hz, 1H, H-4'), 3.73 (dd,** *J***=2.2, 11.6 Hz, 1H, H-7'), 3.69 (m, 1H, H-5), 3.64 (d,** *J***=3.3 Hz, 1H, H-3'), 3.59 (dd,** *J***=5.8, 11.6 Hz, 1H, H-7'), 3.56 (m, 1H, H-4), 3.55 (dd,** *J***=6.6, 8.5 Hz, 1H, H-6), 3.53 (dd,** *J***=9.6, 9.9 Hz, 1H, H-5') 1.39 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): \delta 174.8, 104.4, 81.2, 75.3, 75.1, 74.5, 73.5, 72.6, 70.3, 67.8, 63.6, 62.8, 21.9.**

4.3.9. 1-Deoxy-\alpha-D-manno-hept-2-ulosyl-($2 \rightarrow 2$)-D-mannono-1,5-lactone (16cc). Colorless amorphous solid; ¹H NMR (600 MHz, CD₃OD): δ 4.79 (d, *J*=4.4 Hz, 1H, H-2), 4.43 (dd, *J*=4.4, 2.8 Hz, 1H, H-3), 4.24 (dd, *J*=2.8, 9.1 Hz, 1H, H-4), 3.96 (dd, *J*=3.3, 9.6 Hz, 1H, H-4'), 3.89–3.85 (m, 2H, H-5, H-6'), 3.74 (dd, *J*=2.2, 11.6 Hz, 1H, H-7'), 3.71 (d, *J*=3.3 Hz, 1H, H-3'), 3.7 (dd, *J*=2.6, 11.6 Hz, 1H, H-6), 3.57 (dd, *J*=6.6, 11.6 Hz, 1H, H-7'), 3.56 (dd, *J*=5.2, 11.6 Hz, 1H, H-6), 3.47 (dd, *J*=9.6, 9.6 Hz, 1H, H-5'), 1.39 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): δ 176.0,

103.9, 79.7, 75.6, 74.2, 72.5, 71.4, 70.9, 69.5, 68.1, 64.3, 63.1, 21.9.

4.3.10. 1-Deoxy- α -**D**-*gluco*-hept-2-ulosyl-(2 \rightarrow 3)-**D**-glucono-1,5-lactone (15aa). Colorless amorphous solid; $[\alpha]_{28}^{28}$ +87.7 (*c* 1.02, CH₃OH); ¹H NMR (600 MHz, CD₃OD): δ 4.57 (dd, *J*=6.9, 6.9 Hz, 1H, H-3), 4.49 (d, *J*=6.9 Hz, 1H, H-2), 4.43 (dd, *J*=6.9, 8.0 Hz, 1H, H-4), 3.89 (m, 1H, H-5), 3.72–3.66 (m, 3H, H-6, H-5', H-6'), 3.62–3.53 (m, 3H, H-6, H-4', H-7'), 3.29 (m, 1H, H-7'), 3.14 (d, *J*=9.4 Hz, 1H, H-3'), 1.46 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): δ 176.1, 102.9, 78.8, 77.7, 76.1, 75.3, 74.5, 72.1, 71.4, 71.2, 64.0, 62.2, 21.9.

4.3.11. 1-Deoxy-α-D-*gluco*-hept-2-ulosyl-($2 \rightarrow 3$)-D-galactono-1,5-lactone (15ab). Colorless amorphous solid; $[\alpha]_{25}^{25}$ +44.4 (*c* 1.02, CH₃OH); ¹H NMR (600 MHz, CD₃OD): δ 4.52–4.48 (m, 2H), 4.28 (m, 1H), 3.80 (m, 1H), 3.72–3.69 (m, 2H), 3.60–3.50 (m, 4H), 3.17 (dd, *J*=9.4, 9.6 Hz, 1H), 3.10 (dd, *J*=9.6 Hz, 1H, H-2), 1.45 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): δ 175.9, 102.8, 80.1, 78.2, 76.0, 75.1, 75.1, 74.6, 72.1, 70.5, 63.7, 62.7, 22.1.

4.3.12. 1-Deoxy- α -**D**-*galacto*-hept-2-ulosyl-(2 \rightarrow 3)-D-glucono-1,5-lactone (15ba). Colorless amorphous solid; ¹H NMR (600 MHz, CD₃OD): δ 4.58 (dd, *J*=6.9, 7.2 Hz, 1H, H-3), 4.51 (d, *J*=7.2 Hz, 1H, H-2), 4.43 (dd, *J*=6.9, 7.7 Hz, 1H, H-4), 3.95 (m, 1H), 3.87 (m, 1H, H-5), 3.82 (m, 1H), 3.68–3.65 (m, 2H), 3.63–3.52 (m, 4H), 1.46 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): δ 176.1, 103.2, 78.7, 76.0, 74.4, 73.1, 72.1, 71.9, 71.6, 71.0, 64.0, 62.6, 21.9.

4.3.13. 1-Deoxy-D-galacto-hept-2-ulosyl-($2 \rightarrow 3$)-D-galactono-1,5-lactone (15bb). Colorless amorphous solid; ¹H NMR (600 MHz, CD₃OD): δ 4.53 (d, J=8.0 Hz, 1H, H-2), 4.48 (dd, J=8.0, 8.0 Hz, 1H), 4.27 (dd, J=2.5, 8.0 Hz, 1H, H-3), 3.93 (m, 1H), 3.79 (m, 1H, H-4), 3.71–3.69 (m, 2H), 3.66 (dd, J=3.0, 10.2 Hz, 1H, H-4'), 3.60–3.54 (m, 3H), 3.50 (dd, J=10.2 Hz, 1H, H-3'), 1.48 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): δ 176.0, 103.1, 80.0, 76.0, 74.9, 74.5, 73.7, 71.7, 71.4, 70.6, 63.7, 63.1, 21.9.

4.3.14. 1-Deoxy- α -**D**-*manno*-hept-2-ulosyl-(2 \rightarrow 3)-D-glucono-1,5-lactone (15ca). Colorless amorphous solid; $[\alpha]_{26}^{26}$ +59.6 (*c* 0.69, CH₃OH); ¹H NMR (600 MHz, CD₃OD): δ 4.63 (d, *J*=7.2 Hz, 1H, H-2), 4.59 (dd, *J*=6.9, 7.2 Hz, 1H, H-3), 4.53 (dd, *J*=4.4, 6.9 Hz, 1H, H-4), 3.87 (m, 1H), 3.78 (dd, *J*=3.3, 9.6 Hz, 1H, H-4'), 3.72–3.69 (m, 2H), 3.64–3.58 (m, 3H), 3.59 (d, *J*=3.3 Hz, 1H, H-3'), 3.52 (dd, *J*=9.6, 9.6 Hz, 1H, H-5'), 1.42 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): δ 176.8, 104.1, 80.1, 76.2, 75.2, 74.7, 73.0, 72.7, 72.3, 67.9, 64.2, 62.6, 20.8.

4.3.15. 1-Deoxy- α -**D**-*manno*-hept-2-ulosyl-(2 \rightarrow 3)-D-galactono-1,5-lactone (15cb). Colorless amorphous solid; $[\alpha]_{D}^{25}$ +7.39 (*c* 0.57, CH₃OH); ¹H NMR (600 MHz, CD₃OD): δ 4.61 (dd, *J*=7.7, 8.0 Hz, 1H, H-5'), 4.51 (d, *J*=8.0 Hz, 1H, H-2), 4.23 (dd, *J*=1.7, 8.0 Hz, 1H, H-3), 3.80 (dd, *J*=2.2, 11.3 Hz, 1H, H-6), 3.75 (dd, *J*=3.3, 9.6 Hz, 1H, H-7'), 3.67 (dd, *J*=2.2, 7.7 Hz, 1H, H-4'), 3.67 (m, 1H, H-5), 3.57–3.52 (m, 4H, H-4, H-6, H-3', H-6'), 3.47 (dd, *J*=9.6, 9.9 Hz, 1H, H-7'), 1.44 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): δ 175.9, 104.0, 80.0, 79.6, 75.9, 75.6, 75.0, 74.7, 72.6, 69.7, 68.0, 63.5, 20.8.

4.3.16. 1-Deoxy-\alpha-D-*manno***-hept-2-ulosyl-(2 \rightarrow 3)-D-mannono-1,5-lactone (15cc). Colorless amorphous solid; [\alpha]_{28}^{28} +76.4 (***c* **0.99, CH₃OH); ¹H NMR (600 MHz, CD₃OD): \delta 4.61 (dd,** *J***=3.3 Hz, 1H, H-2), 4.18 (dd,** *J***=3.3, 0.6 Hz, 1H, H-3), 4.15 (ddd,** *J***=2.5, 5.0, 9.9 Hz, 1H, H-6'), 4.03 (ddd,** *J***=2.5, 5.2, 8.0 Hz, 1H, H-5), 3.82 (m, 1H, H-4), 3.75 (dd,** *J***=2.5, 12.7 Hz, 1H, H-6), 3.72 (dd,** *J***=3.3, 9.6 Hz, 1H, H-4'), 3.67 (dd,** *J***=2.5, 11.8 Hz, 1H, H-7'), 3.62 (dd,** *J***=5.0, 12.7 Hz, 1H, H-6), 3.60 (dd,** *J***=5.2, 11.8 Hz, 1H, H-7'), 3.48 (d,** *J***=3.3 Hz, 1H, H-3'), 3.46 (dd,** *J***=9.6, 9.9 Hz, 1H, H-5'), 1.41 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): \delta 175.0, 103.3, 82.6, 76.5, 75.2, 74.4, 72.4, 70.2, 68.8, 68.0, 62.9, 62.4, 22.4.**

4.3.17. 1-Deoxy-\alpha-D-*gluco***-hept-2-ulosyl-(2 \rightarrow 4)-D-glucono-1,5-lactone (20aa). Colorless amorphous solid; [\alpha]_{25}^{25} +93.0 (***c* **0.96, CH₃OH); ¹H NMR (600 MHz, D₂O): \delta 4.65 (m, 1H), 4.41 (m, 1H), 4.15 (m, 1H), 3.74–3.62 (m, 5H), 3.57–3.51 (m, 2H), 3.19 (m, 1H), 3.12 (m, 1H), 1.28 (s, 3H, H-1'); ¹³C NMR (150 MHz, D₂O): \delta 178.0, 102.9, 80.1, 77.0, 74.1, 74.0, 73.8, 73.7, 73.4, 70.7, 62.3, 61.4, 21.4.**

4.3.18. 1-Deoxy-α-D-*gluco*-hept-2-ulosyl- $(2 \rightarrow 4)$ -D-galactono-1,5-lactone (20ab). Colorless amorphous solid; $[\alpha]_D^{25}$ +135.2 (*c* 0.6, H₂O); ¹H NMR (600 MHz, D₂O): δ 4.46 (m, 2H), 4.40 (m, 1H), 4.04 (m, 1H), 3.84 (m, 1H), 3.70 (m, 1H), 3.66–3.58 (m, 3H), 3.54 (m, 1H), 3.24 (m, 1H), 3.19 (m, 1H), 1.28 (s, 3H, H-1'); ¹³C NMR (150 MHz, D₂O): δ 177.0, 102.0, 81.0, 77.0, 74.6, 73.6, 73.3, 73.3, 71.4, 70.3, 61.0, 60.5, 21.4.

4.3.19. 1-Deoxy-\alpha-D-*gluco*-hept-2-ulosyl-(2 \rightarrow 4)-D-mannono-1,5-lactone (20ac). Colorless amorphous solid; $[\alpha]_D^{25}$ +135.2 (*c* 0.6, H₂O); ¹H NMR (600 MHz, D₂O): δ 4.65 (d, *J*=2.8 Hz, 1H, H-2), 4.29 (m, 1H, H-5), 4.03 (m, 1H, H-3), 3.93 (m, 1H, H-4), 3.71 (dd, *J*=2.2, 12.4 Hz, 1H, H-7'), 3.63–3.59 (m, 2H, H-6, H-6), 3.55 (dd, *J*=5.0, 12.4 Hz, 1H, H-7'), 3.46 (dd, *J*=9.4, 9.6 Hz, 1H), 3.37 (ddd, *J*=2.2, 5.0, 9.9 Hz, 1H, H-6'), 3.21 (dd, *J*=9.6, 9.6 Hz, 1H), 3.10 (d, *J*=9.9 Hz, 1H, H-3'), 1.37 (s, 3H, H-1'); ¹³C NMR (150 MHz, D₂O): δ 175.2, 102.7, 81.3, 76.4, 73.9, 73.8, 73.4, 70.3, 70.1, 68.9, 61.9, 61.1, 22.1.

4.3.20. 1-Deoxy- α -**D**-*galacto*-**hept-2-ulosyl-**($2 \rightarrow 4$)-**D**-*g*lucono-1,5-lactone (20ba). Colorless amorphous solid; ¹H NMR (600 MHz, CD₃OD): δ 4.62 (dd, *J*=4.9, 7.1 Hz, 1H, H-4), 4.24 (dd, *J*=3.6, 4.9 Hz, 1H, H-3), 4.21 (d, *J*=3.6 Hz, 1H, H-2), 4.07 (ddd, *J*=1.1, 4.1, 8.2 Hz, 1H, H-6'), 4.05 (ddd, *J*=7.1, 3.3, 3.6 Hz, 1H, H-5), 3.88 (dd, *J*=3.3, 12.7 Hz, 1H, H-6), 3.76 (dd, *J*=3.3, 1.1 Hz, 1H, H-5'), 3.67–3.64 (m, 3H, H-6, H-4', H-7'), 3.56 (dd, *J*=4.1, 11.3 Hz, 1H, H-7'), 3.48 (d, *J*=9.9 Hz, 1H, H-3'), 1.44 (s, 3H, H-1').

4.3.21. 1-Deoxy- α -**D**-*galacto*-**hept-2-ulosyl-**(**2** \rightarrow **4**)-**D**-**galactono-1,5-lactone** (**20bb**). Colorless amorphous solid; ¹H NMR (600 MHz, CD₃OD): δ 4.99 (dd, *J*=2.2, 5.8 Hz, 1H, H-4), 4.33 (dd, *J*=5.8, 6.3 Hz, 1H, H-3), 4.17 (d, *J*=6.3 Hz, 1H, H-2), 3.96–3.90 (m, 2H), 3.78–3.76 (m, 2H), 3.62–3.55 (m, 4H), 3.46 (d, *J*=9.9 Hz, 1H), 1.46 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): δ 177.1, 102.9, 83.0, 76.3, 75.4, 75.3, 73.4, 73.2, 71.3, 71.1, 62.7, 61.5, 21.6.

4.3.22. 1-Deoxy-α-D-*galacto*-hept-2-ulosyl- $(2 \rightarrow 4)$ -D-mannono-1,5-lactone (20bc). Colorless amorphous solid; ¹³C NMR (150 MHz, CD₃OD): δ 174.8, 103.5, 82.9, 74.7, 74.6, 73.7, 71.6, 71.1, 70.8, 69.6, 63.2, 62.8, 22.9.

4.3.23. 1-Deoxy-α-D*manno***-hept-2-ulosyl-**($2 \rightarrow 4$)**-D-galactono-1,5-lactone** (**20cb**). Colorless amorphous solid; ¹H NMR (600 MHz, CD₃OD): δ 4.37 (m, 2H), 4.18 (dd, J=8.5, 8.5 Hz, 1H), 4.13 (m, 1H), 4.02 (dd, J=5.0, 11.3 Hz, 1H), 3.84–3.80 (m, 2H), 3.71–3.62 (m, 4H), 3.52 (m, 1H), 1.46 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): δ 177.1, 102.9, 83.0, 76.3, 75.4, 75.3, 73.4, 73.2, 71.3, 71.1, 62.7, 61.5, 21.6.

4.3.24. 1-Deoxy- α -**D**-*manno*-**hept-2-ulosyl-(2** \rightarrow **4**)-**D**-mannono-1,5-lactone (20cc). Colorless amorphous solid; $[\alpha]_{25}^{25}$ +76.8 (*c* 0.48, CH₃OH); ¹H NMR (600 MHz, CD₃OD): δ 4.47 (d, *J*=3.0 Hz, 1H, H-2), 4.34 (ddd, *J*=5.2, 5.5, 3.3 Hz, 1H, H-5), 4.12 (dd, *J*=2.2, 5.2 Hz, 1H, H-4), 4.01 (ddd, *J*=3.0, 2.2 Hz, 1H, H-3), 3.78 (dd, *J*=3.3, 12.4 Hz, 1H, H-6), 3.74 (dd, *J*=2.2, 11.8 Hz, 1H, H-7'), 3.73–3.69 (m, 2H, H-4', H-6), 3.59 (dd, *J*=6.1, 11.8 Hz, 1H, H-7'), 3.54 (d, *J*=3.3 Hz, 1H, H-3'), 3.48 (dd, *J*=9.6, 9.9 Hz, 1H, H-5'), 3.39 (ddd, *J*=2.2, 6.1, 9.9 Hz, 1H, H-6'), 1.45 (s, 3H, H-1'); ¹³C NMR (150 MHz, CD₃OD): δ 174.4, 104.4, 83.7, 76.2, 75.1, 74.8, 72.5, 70.5, 69.5, 67.9, 63.4, 63.0, 21.7.

4.3.25. 1-Deoxy- α -**D**-*galacto*-hept-2-ulosyl-(2 \rightarrow 6)-**D**-galactono-1,5-lactone (21bb). Colorless amorphous solid; ¹H NMR (600 MHz, CD₃OD): δ 4.26 (d, J=8.8 Hz, 1H, H-2), 4.17 (dd, J=8.3, 8.8 Hz, 1H, H-3), 4.08 (dd, J=3.3, 8.3 Hz, 1H, H-4), 3.80–3.78 (m, 2H, H-5, H-5'), 3.70 (dd, J=3.3, 9.9 Hz, 1H, H-4'), 3.68 (m, 1H, H-6'), 3.61 (dd, J=6.9, 11.3 Hz, 1H), 3.58 (dd, J=5.5, 11.3 Hz, 1H), 3.55 (dd, J=5.8, 9.6 Hz, 1H), 3.51 (dd, J=6.3, 9.6 Hz, 1H), 3.47 (dd, J=9.9 Hz, 1H, H-3'), 1.34 (s, 3H, H-1').

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Supplementary data

Supplementary data associated with this article can be found in the online version, at doi:10.1016/j.tet.2006.07.032.

References and notes

- Boons, G.-J.; Demchenko, A. V. *Carbohydrate-based Drug Discovery*; Wong, C. H., Ed.; Wiley-VCH: Weinheim, 2003; Vol. 1, pp 55–102.
- Johnson, L. N.; Hu, S.-H.; Barford, D. Faraday Discuss. 1992, 93, 131–142.
- For recent reviews on sialylation: (a) Boons, G.-J.; Demchenko, A. V. *Chem. Rev.* 2000, 100, 4539–4565; (b) Hansson, J.; Oscarson, S. *Curr. Org. Chem.* 2000, 4, 535– 564; (c) Kiefel, M. J.; Itzstein, M. *Chem. Rev.* 2002, 102, 471–490.

- 4. Krog-Jensen, C.; Oscarson, S. J. Org. Chem. 1998, 63, 1780-1784.
- 5. Dondoni, A. *Pure Appl. Chem.* **2000**, *72*, 1577–1588 and references cited therein.
- (a) Li, X. L.; Ohtake, H.; Takahashi, H.; Ikegami, S. *Tetrahedron* 2001, 57, 4283–4295; (b) Li, X. L.; Ohtake, H.; Takahashi, H.; Ikegami, S. *Synlett* 2001, 1885–1888; (c) Namme, R.; Mitsugi, T.; Takahashi, H.; Ikegami, S. *Tetrahedron Lett.* 2005, 46, 3033–3036.
- (a) Li, X. L.; Ohtake, H.; Takahashi, H.; Ikegami, S. *Tetrahedron* 2001, *57*, 4297–4309; (b) Li, X. L.; Takahashi, H.; Ohtake, H.; Shiro, M.; Ikegami, S. *Tetrahedron* 2001, *57*, 8053–8066.
- Yamanoi, T.; Oda, Y.; Yamazaki, I.; Shinbara, M.; Morimoto, K.; Matsuda, S. *Lett. Org. Chem.* 2005, *2*, 242–246.
- Other groups also investigated the O-glycosidation of *exo*-glycals: (a) Chang, C.-F.; Yang, W.-B.; Chang, C.-C.; Lin, C.-H. *Tetrahedron Lett.* 2002, 43, 6515–6519; (b) Lin, H.-C.; Yang, W.-B.; Gu, Y.-F.; Chen, C.-Y.; Wu, C.-Y.; Lin, C.-H. Org. Lett. 2003, 5, 1087–1089; (c) Colinas, P. A.; Lieberknecht, A.; Bravo, R. D. *Tetrahedron Lett.* 2002, 43, 9065–9068; (d) Colinas, P. A.; Ponzinibbio, A.; Lieberknecht, A.; Bravo, R. D. *Tetrahedron Lett.* 2003, 44, 7985–7988.
- Enzymatic O-glycosidation of heptenitols: Schlesselmann, P.; Fritz, H.; Lehmann, J.; Uchiyama, T.; Brewer, C. F.; Hehre, E. J. *Biochemistry* 1982, *21*, 6606–6614.
- O-Glycosidation of heptenitols promoted by iodonium ion, see: Noort, D.; Veeneman, G. H.; Boons, G.-J. P. H.; van der Marel, G. A.; Mulder, G. J.; van Boom, J. H. *Synlett* **1990**, 205–206.
- For reviews of *exo*-glycal chemistry, see: (a) Taillefumier, C.; Chapleur, Y. *Chem. Rev.* **2004**, *104*, 263–292; (b) Lin, C.-H.; Lin, H.-C.; Yang, W.-B. *Curr. Top. Med. Chem.* **2005**, *5*, 1431–1457.
- 13. (a) RajanBabu, T. V.; Reddy, G. S. J. Org. Chem. 1986, 51, 5458–5461; (b) Petasis, N. A.; Lu, S.-P. Tetrahedron Lett. 1995, 36, 2393–2396.
- 14. For synthetic procedures of compounds **7–10**, see Supplementary data.
- 15. CCDC 610513 contains the supplementary crystallographic data for compound **14cc**. These data can be obtained free of charge from The Cambridge Crystallographic Data Center via www.ccdc.cam.ac.uk/data_request/cif.
- Tvaroska, I.; Travel, F. R. Carbon–proton coupling constants in the conformational analysis of sugar molecules. In *Advances in Carbohydrate Chemistry and Biochemistry*; Horton, D., Ed.; Academic: San Diego, CA, 1995; Vol. 51, pp 15–61.
- 17. Bieg, T.; Szeja, W. Synthesis 1985, 76-77.
- 18. Liu, H.-J.; Yip, J. Tetrahedron Lett. 1997, 38, 2253-2256.
- Schuda, P. F.; Cichowicz, M. B.; Heimann, M. R. *Tetrahedron Lett.* **1983**, *24*, 3829–3830.
- 20. Binkley, R. W.; Hehmann, D. G. J. Org. Chem. 1990, 55, 378–380.
- (a) Hirama, M.; Shimizu, M. Synth. Commun. 1983, 13, 781– 786; (b) Angibeaud, P.; Defaye, J.; Gadelle, A.; Utille, J.-P. Synthesis 1985, 1123–1125.
- (a) Czech, B. P.; Bartsh, R. A. J. Org. Chem. 1984, 49, 4076–4078; (b) Sajiki, H.; Hirota, K. Tetrahedron 1998, 54, 13981–13996.
- 23. In some cases, decomposition of debenzylated disaccharides 15, 16, 20, and 21 was observed during work-up. Filtration through a celite pad caused decomposition, and therefore filter paper was used for separation of the catalyst.
- 24. For the peak assignments in ¹H and ¹³C NMR spectra for compounds **11–14**, see Supplementary data.